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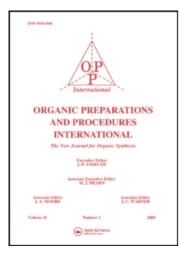
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PREPARATION OF 1-(o-AMINOBENZOYL)-1-METHYLHYDRAZINES

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2-Aminobenzoylhydrazines are important starting compounds for the synthesis of various heterocycles. With oxo reagents for example, quinazolines, oxadiazoles or benzotriazepinones are formed depending on the reagent and conditions used.¹ The easiest method for the preparation of 2-aminobenzoylhydrazines is the treatment of isatoic anhydrides² with hydrazines. When monoaryl or acetylated hydrazines are used, 2-(o-aminobenzoyl)-1-substituted hydrazines (3) are formed.^{2,3} With methylhydrazine, the main product is of type 2 since the methyl group increases the basicity of the substituted nitrogen. Sunder et al.⁴ reported that the ring opening of isatoic anhydride with

$$X \longrightarrow NH_2NHMe$$

NH₂

methylhydrazine in dry DMF resulted in an 85:15 mixture of 2a and 3a, whereas Leiby and Heindel⁵ stated that the same reaction and the ring opening of 5-chloroisatoic anhydride (1b) in 95% ethanol resulted in 2b and 3b, respectively, as the sole products. Both groups of authors described the products as oils which were used in different reactions.⁶ When these experiments were repeated under the reported conditions, the ¹H NMR (400 MHz) spectra showed the distilled oily products to be 9:1 mixtures of 2:3. When diethyl ether was added to these oils, 2 crystallized (2b rapidly and 2a more slowly). After two recrystallizations, the isomer-free 2a and 2b were obtained. From the mother liquor of 2a, the minor product 3a was separated by means of TLC. The products were stable at room temperature.

EXPERIMENTAL SECTION

Melting points were determined in capillary tubes with an Electrothermal Digital Melting Point apparatus and are uncorrected. ¹H NMR spectra were recorded on a JEOL GX 400 FT NMR spectrometer in CDCl₃ with TMS as internal standard. MS spectra were obtained on an MM 7070E mass spectrometer at 70 eV.

Reaction of Isatoic Anhydrides (1a or 1b) with Methylhydrazine. To a slurry of isatoic anhydride (0.1 mol) in ethanol (100 ml), methylhydrazine (6 g, 0.13 mol) was added over a 5 min period. A

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clear solution was obtained with concurrent evolution of carbon dioxide. After stirring for 4 hrs, the solvent was evaporated and the residue was distilled at 170-185° (1 mm) to afford a 9:1 mixture of 2:3. The oily product was dissolved in ether (100 ml) and left to stand overnight at 4°. The isomerfree product (2a or 2b) was collected and recrystallized.

2a: 10.0 g, (60%), mp. 71-73° (3:1 diisopropyl ether-ethanol). ¹H NMR (CDCl₃): δ 3.24 (s, 3H, CH₃), 4.50 (s, 4H, 2 x NH₂), 6.70-6.74 (m, 2H, Ar), 7.13-7.20 (m, 2H, Ar). MS (EI): m/z(%) 165(22), 120(100), 92(32), 65(7).

<u>Anal</u>. Calcd for C₈H₁₁N₃O; C, 58.16; H, 6.71; N, 25.44. Found: C, 58.24; H, 6.86; N, 25.39

2b: 13.2 g (66%), mp. 82-84° (9:1 diisopropyl ether-ethanol). ¹H NMR (CDCl₃): δ 3.25 (s, 3H, CH₃), 4.47 (s, 4H, 2 x NH₂), 6.65-6.67 (m, 1H, Ar), 7.13-7.20 (m, 2H, Ar). MS (EI): m/z (%) 199(24), 154(100), 126(26), 99(12).

<u>Anal.</u> Calcd for C₈H₁₀ClN₃O: C, 48.13; H, 5.05; N, 21.05. Found: C, 48.20; H, 5.12; N, 20.96

From the mother liquor obtained from the purification of 2a, separation on a preparative TLC plate (Merck) with 4:1 benzene-ethanol gave small amounts of 3a as white crystals, mp. 90-91° (hexane), lit.⁶ mp. 90-91°. ¹H NMR (CDCl₃): δ 2.71 (s, 3H, CH₃), 5.47 (s, 3H, NH and NH₂), 6.63-6.70 (m, 2H, Ar), 7.20-7.30 (m, 2H, Ar), 7.60 (s, 1H, NHCH₃). MS(EI): m/z(%) 165(22), 136(5), 120(100), 92(37), 65(28).

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